

Redox Practice Solutions.pdf

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Additional Practice Problems Acidic Solution 1. $\text{Ag} + \text{NO}_3^- \rightarrow \text{Ag}^+ + \text{NO}$ Answer: ... Microsoft Word - Oxidation-Reduction Extra Practice.doc Author: aallan Created Date: 10/13/2011 12:51:26 PM ...

Practice Problems: Redox Reactions

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Practice Problems: Redox Reactions (Answer Key) - Patti Lab

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Practice Problems: Redox Reactions (Answer Key) ... Write balanced equations for the following redox reactions: a. $2 \text{NaBr} + \text{Cl}_2 \rightarrow 2 \text{NaCl} + \text{Br}_2$ b. $\text{Fe}_2\text{O}_3 + 3 \text{CO} \rightarrow 2 \text{Fe} + 3 \text{CO}_2$ in acidic solution c. $5 \text{CO} + \text{I}_2\text{O}_5 \rightarrow 5 \text{CO}_2 + \text{I}_2$ in basic solution ; ... Return to Practice Problems Page ...

Balancing redox reactions in acidic solution: Problems #1-10

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Balancing redox reactions in acidic solution Problems #1-10 Examples 1-5 Only the examples and problems; Problems 11-30 Balancing in basic solution ... You cannot have electrons appear in the final answer of a redox reaction. (You can in a half-reaction, but remember half-reactions do not occur alone, they occur in reduction-oxidation pairs.) ...

Balancing REDOX Reactions: Learn and Practice

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Balancing REDOX Reactions: Learn and Practice Reduction-Oxidation reactions (or REDOX reactions) occur when the chemical species involved ... Now, we must consider more complicated REDOX reactions. Aqueous solutions in electrochemistry are often acidic or basic. The protons (H^+) and hydroxides (OH^-) contribute the

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